Acta Crystallographica Section E

## Structure Reports

Online
ISSN 1600-5368

## Bin Tan, ${ }^{\text {a }}$ Jian-Feng Zheng, ${ }^{a}$ Xiang-Jiang Kong ${ }^{\text {b }}$ and Li-Ren Jin ${ }^{\text {a }}$

${ }^{\text {a }}$ The Key Laboratory for Chemical Biology of Fujian Province, Department of Chemistry, Xiamen University, Xiamen 361005, People's Republic of China, and ${ }^{\mathbf{b}}$ Department of
Chemistry, Xiamen University, Xiamen 361005, People's Republic of China

Correspondence e-mail: Irjin@xmu.edu.cn

## Key indicators

Single-crystal X-ray study
$T=273 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.008 \AA$
$R$ factor $=0.072$
$w R$ factor $=0.215$
Data-to-parameter ratio $=16.0$
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.
© 2006 International Union of Crystallography Printed in Great Britain - all rights reserved

## (3S)-4-[(1S)-1-(Dibenzylamino)-2-phenyl-ethyl]-1,3,2-dioxathiane

The title compound, $\mathrm{C}_{25} \mathrm{H}_{27} \mathrm{NO}_{3} \mathrm{~S}$, was obtained as the major product from the reaction of ( $3 S, 4 S$ )-4-(dibenzylamino)-5-phenylpentane-1,3-diol and sulfuryl dichloride. The molecular packing in the crystal is stabilized by weak intermolecular interactions and van der Waals forces.

## Comment

The title compound, (I), is an important intermediate in the synthesis of a chiral tridentate ligand. Single-crystal analysis of (I) facilitated the characterization of the compound with two stereomeric centers. The heterocyclic ring is in chair form, with the alkyl chain attached in the equatorial position.

(I)

## Experimental

To a solution of ( $3 S, 4 S$ )-4-(dibenzylamino)-5-phenylpentane-1,3-diol ( $750 \mathrm{mg}, 2 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{C1}_{2}(15 \mathrm{ml})$ and diethyl ether ( 5 ml ) were added triethylamine ( $1.4 \mathrm{ml}, 10 \mathrm{mmol}$ ) and then thionyl chloride


Figure 1
ORTEP 3 (Farrugia, 1997) plot of (I), with displacement ellipsoids drawn at the $50 \%$ probability level. H atoms are drawn as spheres of arbitrary radii.

Received 16 December 2005 Accepted 19 December 2005 Online 7 January 2006

## organic papers

$(0.4 \mathrm{ml}, 5.4 \mathrm{mmol})$ at 273 K . The resulting solution was stirred for 1 h (Van Woerden, 1963). The mixture was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(20 \mathrm{ml})$ and water ( 10 ml ), the organic phase was washed with brine ( 8 ml ) and dried over $\mathrm{MgSO}_{4}$ and concentrated to give the crude product; flash chromatographic purification on silica gel yielded the product ( 515 mg , yield: $61 \%$ ). Suitable crystals were obtained by crystallization from 2-propanol and petroleum ether ( $1: 3 \mathrm{v} / \mathrm{v}$ ). MS (ESI): $422.4\left(\mathrm{MH}^{+}\right)$. IR (film): 3063, 3025, 2930, 2806, 1491, 1449, 1192, 969, $869,694 \mathrm{~cm}^{-1.1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 500 \mathrm{MHz}\right): \delta 0.92(d, J=14.0 \mathrm{~Hz}, 1 \mathrm{H}$, $\left.\mathrm{CH} a \mathrm{H} b \mathrm{CH}_{2} \mathrm{O}\right), 2.73\left(m, 1 \mathrm{H}, \mathrm{CH} a \mathrm{H} b \mathrm{CH}_{2} \mathrm{O}\right), 2.90(m, 2 \mathrm{H}, \mathrm{PhCH} a \mathrm{H} b$, CHN), 3.04 ( $m, 1 \mathrm{H}, \mathrm{PhCH} a \mathrm{H} b$ ), 3.49 ( $d d, J=5.0,13.0 \mathrm{~Hz}, 2 \mathrm{H}$, $2 \mathrm{PhCH} a \mathrm{H} b \mathrm{~N}), 3.88(m, 1 \mathrm{H}, \mathrm{CH} a \mathrm{H} b \mathrm{O}), 4.13(d, J=13.0 \mathrm{~Hz}, 2 \mathrm{H}$, $2 \mathrm{PhCH} a \mathrm{H} b \mathrm{~N}), 4.80(m, 1 \mathrm{H}, \mathrm{CH} a \mathrm{H} b \mathrm{O}), 4.84(d, J=12.0 \mathrm{~Hz}, 1 \mathrm{H}$, CHO), 7.13-7.38 ( $m, 15 \mathrm{H}, \mathrm{PhH}$ ). ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 125 \mathrm{MHz}\right): \delta$ 28.69, 29.16, 54.43, 58.23, 61.89, 68.20, 125.97, 126.92, 128.22, 128.30, 129.23, 139.23, 139.58

## Crystal data

| $\mathrm{C}_{25} \mathrm{H}_{2} \mathrm{NO}_{3} \mathrm{~S}$ | Mo $K \alpha$ radiation <br> $M_{r}=421.54$ |
| :--- | :--- |
| Orthorhombic, $P 2_{1} 2_{1} 2_{1}$ | reflectioteters from 5908 |
| $a=10.855(2) \AA \AA$ | $\theta=2.2-22.0^{\circ}$ |
| $b=11.643(2) \AA$ | $\mu=0.17 \mathrm{~mm}^{-1}$ |
| $c=18.002(4) \AA$ | $T=273(2) \mathrm{K}$ |
| $V=2275.1(8) \AA^{3}$ | Chunk , colourless |
| $Z=4$ | $0.48 \times 0.36 \times 0.28 \mathrm{~mm}$ |
| $D_{x}=1.231 \mathrm{Mg} \mathrm{m}$ |  |
| Data collection |  |
| Bruker APEX area-detector |  |
| diffractometer | 4004 independent reflections |
| $\varphi$ and $\omega$ scans | 3426 reflections with $I>2 \sigma(I)$ |
| Absorption correction: multi-scan | $R_{\text {int }}=0.025$ |
| (SADABS; Bruker, 2001) | $\theta_{\text {max }}=25.0^{\circ}$ |
| $T_{\text {min }}=0.924, T_{\text {max }}=0.955$ | $h=-12 \rightarrow 12$ |
| 16447 measured reflections | $k=-13 \rightarrow 13$ |
|  | $l=-20 \rightarrow 21$ |

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.072$
$w R\left(F^{2}\right)=0.215$
$S=1.05$
4004 reflections
251 parameters
H -atom parameters constrained

$$
\begin{aligned}
& w=1 /\left[\sigma^{2}\left(F_{\mathrm{o}}^{2}\right)+(0.1146 P)^{2}\right. \\
& \quad+0.7614 P] \\
& \text { where } P=\left(F_{\mathrm{o}}^{2}+2 F_{\mathrm{c}}^{2}\right) / 3 \\
& (\Delta / \sigma)_{\max }=0.003 \\
& \Delta \rho_{\max }=0.42 \mathrm{e} \AA^{-3} \\
& \Delta \rho_{\min }=-0.22 \mathrm{e} \AA^{-3} \\
& \text { Absolute structure: Flack (1983), } \\
& 1720 \text { Friedel pairs } \\
& \text { Flack parameter: } 0.2(2)
\end{aligned}
$$

H atoms were positioned geometrically $(\mathrm{C}-\mathrm{H}=0.93,0.93$ and $0.96 \AA$ for phenyl, methine and methylene H atoms, respectively) and refined as riding, with $U_{\text {iso }}(\mathrm{H})=1.2 U_{\text {eq }}$ (parent atom).

Data collection: SMART (Bruker, 2001); cell refinement: SAINT (Bruker, 2001); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 (Farrugia, 1997) and ViewerPro (Accelrys, 2001); software used to prepare material for publication: SHELXL97.

The authors thank the Fujian Science Foundation and Xiamen Science Foundation for financial support. We also thank the Key Laboratory for Physical Chemistry of the Solid Surface for providing the X-ray diffraction facilities.

## References

Accelrys (2001). ViewerPro. Version 4.2. Accelrys Inc., Burlington, Massachusetts, USA.
Bruker (2001). SAINT (Version 6.22), SMART (Version 5.625) and SADABS (Version 2.03). Bruker AXS Inc., Madison, Wisconsin, USA.
Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.
Flack, H. D. (1983). Acta Cryst. A39, 876-881.
Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
Van Woerden, H. F. (1963). Chem. Rev. 63, 557-571.

